CHEMETALL FOOTE CORPORATION

MATERIAL SAFETY DATA SHEET

LITHIUM CARBONATE CFM 051 Page 1 of 10

SECTION 1 CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

CHEMETALL FOOTE CORPORATION FOR EMERGENCY TRANSPORTATION

348 HOLIDAY INN DRIVE INFORMATION, CALL CHEMTREC

KINGS MOUNTAIN, NC 28086 1-800-424-9300

704-739-2501 (8 AM - 5 PM M-F)

SUBSTANCE: LITHIUM CARBONATE RTECS NUMBER: 0J5800000

TRADE NAMES/SYNONYMS:

CARBONIC ACID, DILITHIUM SALT; DILITHIUM CARBONATE; CARBONIC ACID, LITHIUM SALT; LITHIUM CARBONATE (L12CO); CARBOLITH; ESKALITH; HYPNOREX; LITHONATE; LITHOTABS; PLENUR; L-119; CL12O3; CI-M12880

CHEMICAL FAMILY: Inorganic Salt

CREATION DATE: 05/08/95 REVISION DATE: 10/23/98

SECTION 2 COMPOSITION, INFORMATION ON INGREDIENTS

Component	CAS	% w/w	Exposure Limits in Air				
			ACGIH		OSHA		OTHER
			TLV	STEL	PEL	STEL	1
Lithium Carbonate	554-13-2	> 99	10 mg/m ³ ; Inhalable Particulate; 3 mg/m ³ , Respirable Particulate	NE	5 mg/m ³ ; Respirable fraction 15 mg/m ³ ; Total Dust	NE	NE
			(Particulates not Otherwise Classified)		(Particulates not Otherwise Classified)		

NE = Not Established. C = Ceiling Limit. See Section 16 for Definitions of Terms Used.

NOTE: All WHMIS required information is included. It is located in appropriate sections based on the ANSI Z400.1-1993 format.

SECTION 3 HAZARDS IDENTIFICATION

EMERGENCY OVERVIEW: This is a white, odorless solid. This product can be irritating to contaminated skin and eyes. If inhaled, especially in large quantities, lung, liver and central nervous system effects can occur. This product is not flammable and is not reactive under most circumstances. Emergency responders must wear adequate personal protective equipment for the situations to which they are responding.

SYMPTOMS OF OVER-EXPOSURE BY ROUTE OF EXPOSURE: The most serious health consequence reported for lithium carbonate has been adverse effects on the central nervous system, and liver and thyroid disorders from chronic over-exposure to this compound through ingestion (during medical treatments). In terms of anticipated occupational over-exposure situations for employees, the main health effect from over-exposure would be irritation of contaminated skin and eyes.

<u>INHALATION</u>: Inhalation of dusts may irritate the eyes, nose, and respiratory system. Inhalation of relatively large doses of this product may cause chemical pneumonitis and pulmonary edema. Symptoms of such over-exposure include nausea, headache, coughing, and inflammation of the bronchi.

CONTACT WITH SKIN or EYES: Lithium Carbonate is a severe skin and eye irritant. Over-exposure of the skin can lead to itching, pain, and reddening. Prolonged or repeated skin exposures can lead to dermatitis (inflammation of the outer layer of the skin). Contact with the eyes can cause pain and reddening of the eye tissue.

LITHIUM CARBONATE CFM 051 Page 2 of 10 HAZARDS IDENTIFICATION (Continued) SECTION 3 SKIN ABSORPTION: Skin absorption is not a significant route of exposure for lithium carbonate. INGESTION: Ingestion of large doses can impact the central nervous system, which can produce symptoms which appear as "drunkenness" (i.e. drowsiness, stumbling, dizziness, personality change). Repeated ingestion of this product may cause rash, ringing in the ears, nausea, vomiting, diarrhea, difficulty speaking, drowsiness, twitching, visual disturbances and coma. Ingestion of relatively large quantities of lithium carbonate can result in kidney and thyroid disorders. **INJECTION:** Over-exposure via injection of this product can lead to pain and irritation at the point of injection; additionally, symptoms such as those described for "Ingestion" may develop. HEALTH EFFECTS OR RISKS FROM EXPOSURE: An Explanation in Lay Terms. ACUTE: In terms of occupational use situations, the chief health effect anticipated after over-exposure would be irritation of contaminated skin and eyes. CHRONIC: Allergic dermatitis (cracking and reddening of the skin) may develop after prolonged or repeated skin contact with this product. Long-term over-exposure via inhalation or ingestion can produce kidney and thyroid disorders and central nervous system effects. Lithium carbonate is a reproductive toxin. Refer to Section 11 (Texicological Information) for additional information. HAZARDOUS MATERIAL IDENTIFICATION SYSTEM RATING: Health Hazard = 2; Fire Hazard = 0; Reactivity Hazard Rating = 0; PPE Rating = C **SECTION 4** FIRST-AID MEASURES SKIN EXPOSURE: If this product contaminates the skin, immediately begin decontamination with running water. Remove exposed or contaminated clothing, taking care not to contaminate eyes. Victims must seek immediate medical attention. EYE EXPOSURE: If this product enters the eyes, open victim's eyes while under gentle running water. Use sufficient force to open eyelids. Have victim "roll" eyes. Minimum flushing is for 15 minutes. Victims must seek immediate medical attention. INHALATION: If this product is inhaled, remove victim to fresh air. If necessary, use artificial respiration to support vital functions. Remove or cover gross contamination to avoid exposure to rescuers. INGESTION: If an over-exposure via ingestion occurs, CALL PHYSICIAN OR POISON CONTROL CENTER FOR MOST CURRENT INFORMATION. If professional advice is not available, induce vomiting (only if victim is conscious and is not having convulsions). Victim should drink milk, egg whites, or large quantities of water. Never induce vomiting or give diluents (milk or water) to someone who is unconscious, having convulsions, or who cannot swallow. Victims of chemical exposure must be taken for medical attention. Rescuers should be taken for medical attention, if necessary. Take copy of label and MSDS to physician or health professional with victim. **SECTION 5** FIRE-FIGHTING MEASURES FLASH POINT, °C (method): Not flammable. AUTOIGNITION TEMPERATURE. °C: Not applicable. FLAMMABLE LIMITS (in air by volume): Not applicable FIRE EXTINGUISHING MATERIALS: This product is not flammable. Use fire extinguishing material appropriate for surrounding fires.

Halon: YES

Other: Any "ABC" Class.

Carbon Dioxide: YES

FOATS: YES

Dry Chemical: YES

SECTION 5 FIRE-FIGHTING MEASURES (Continued)

<u>UNUSUAL FIRE AND EXPLOSION HAZARDS</u>: When involved in a fire, this material may decompose and produce irritating fumes and toxic gases (lithium compounds, carbon oxides).

Explosion Sensitivity to Mechanical Impact: Not sensitive. Explosion Sensitivity to Static Discharge: Not sensitive.

<u>SPECIAL FIRE-FIGHTING PROCEDURES</u>: Incipient fire responders should wear eye protection. Structural fire fighters must wear Self-Contained Breathing Apparatus and full protective equipment

NFPA RATING: Health Hazard = 2; Fire Hazard = 0; Reactivity Hazard Rating = 0.

SECTION 6 ACCIDENTAL RELEASE MEASURES

Uncontrolled releases should be responded to by trained personnel using pre-planned procedures. Proper protective equipment should be used. In case of a spill, clear the affected area, protect people, and respond with trained personnel.

The minimum Personal Protective Equipment recommended for response to non-incidental releases should be Level C: triple-gloves (rubber gloves and nitrile gloves, over latex gloves), chemically resistant suit and boots, hard-hat, and air-purifying respirator with high-efficiency particulate filter. Level B with a Self-Contained Breathing Apparatus should be worn in situations where the oxygen level is below 19.5 % or is unknown.

Sweep-up or vacuum spilled material carefully, avoiding the generation of dusts. Decontaminate the area thoroughly. Place all spill residue in an appropriate container. Dispose of in accordance with Federal, State, and local solid waste disposal regulations (see Section 13, Disposal Considerations).

SECTION 7 HANDLING AND STORAGE

WORK PRACTICES AND HYGIENE PRACTICES: As with all chemicals, avoid getting this product ON YOU or IN YOU. Do not eat or drink while handling this product. Wash hands after handling this material. Avoid creating dusts of this product.

STORAGE AND HANDLING PRACTICES: All employees who handle this material should be trained to handle it safely. Avoid breathing dusts or particles generated by this product. Wash thoroughly after using this material. Read instructions provided with the product prior to use.

PROTECTIVE PRACTICES DURING MAINTENANCE OF CONTAMINATED EQUIPMENT: Follow practices indicated in Section 6 (Accidental Release Measures). Make certain application equipment is locked and tagged-out safely, as applicable. Always use this product in areas where adequate ventilation is provided. Decontaminate equipment using soapy water before maintenance begins. Collect all rinsates and dispose of according to applicable Federal, State, or local procedures.

SECTION 8 EXPOSURE CONTROLS, PERSONAL PROTECTION

<u>VENTILATION AND ENGINEERING CONTROLS</u>: Use with adequate ventilation. Mechanical exhaust may be needed. Ensure eyewash/safety shower stations are available near areas where this product is used.

RESPIRATORY PROTECTION: Maintain airborne contaminant concentrations below exposure limits listed in Section 2 (Composition and Information on Ingredients). If respiratory protection is needed, use only protection authorized in 29 CFR 1910.134, or applicable State regulations. Use supplied air respiration protection if oxygen levels are below 19.5% or are unknown.

LITHIUM CARBONATE CFM 051 Page 4 of 10 **SECTION 8** EXPOSURE CONTROLS, PERSONAL PROTECTION (Continued) EYE PROTECTION: Splash goggles or safety glasses. HAND PROTECTION: Wear neoprene gloves for routine industrial use. BODY PROTECTION: Use body protection appropriate for task (i.e. Apron or Tyvek suit). SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES RELATIVE VAPOR DENSITY (air = 1): Not applicable. EVAPORATION RATE: Not applicable. SPECIFIC GRAVITY: 2.1 MELTING/FREEZING POINT: 723 °C, (1333 °F) SOLUBILITY IN WATER: 1.54% @ 20 °C BOILING POINT: 1310 °C, (2390 °F) VAPOR PRESSURE, mm Hg @ 20 °C: 0 pH: 11.2 @ 1% solution ODOR THRESHOLD: Not available. COEFFICIENT WATER/OIL DISTRIBUTION: Not available. APPEARANCE AND COLOR: White, odorless solid. HOW TO DETECT THIS SUBSTANCE (warning properties): This product does not have any unique warning properties. STABILITY AND REACTIVITY SECTION 10 STABILITY: Stable. **DECOMPOSITION PRODUCTS:** Thermal decomposition of the components of this product include toxic oxides of carbon. MATERIALS WITH WHICH SUBSTANCE IS INCOMPATIBLE: This product is not compatible with strong acids and strong oxidizers. HAZARDOUS POLYMERIZATION: Will not occur. CONDITIONS TO AVOID: Avoid mixing this product with incompatible chemicals. **SECTION 11** TOXICOLOGICAL INFORMATION TOXICITY DATA: Additional toxicology information for Lithium Carbonate is provided below: DNA Damage System (human, fibroblast) = 500 mg/L TDLo (oral, man) = 8 mg/kg/gastrointestinal effects/skin effects. TDLo (oral, man) = 54 mg/kg

Mammalian Somatic Cell (hamster, ingestion) = 2 g/L TDLo (oral, woman) = 4256 mg/kg/1-38 weeks pregnant/reproductive effects.

TDLo (oral, woman) = 4900 mg/kg/1-35 weeks pregnant/teratogenic effects TDLo (oral, woman) = 3600 mg/kg/21 weeks continuous/carcinogenic

effects/blood effects

TD (oral, woman) = 5940 mg/kg/47 weeks continuous/carcinogenic effects/blood effects TD (oral, man) = 6132 mg/kg/2 years continuous/carcinogenic effects/blood effects

TD (oral, woman) = 21 g/kg/3.5 years continuous/carcinogenic effects

TDLo (unreported, woman) = 556 mg/kg/32 days. $LD_{50}(oral, rat) = 525 \text{ mg/kg}$

 $LD_{50}(oral, mouse) = 531 mg/kg$ LD_{so}(intraperitoneal, mouse) = 236 mg/kg LDs/(subcutaneous, mouse) = 413 mg/kg LD30(intravenous, mouse) = 497 mg/kg $LD_{30}(oral, dog) = 500 \text{ mg/kg}$

TDLo (oral, woman) = 120 mg/kg/10 days intermittent

TDLo (oral, man) = 1080 mg/kg/13 weeks intermittent/skin effects.

LD₂₀(intraperitoneal, rat) = 156 mg/kg LD₂₀(subcutaneous, rat) = 434 mg/kg LD_{so}(intravenous, rat) = 241 mg/kg

TDLo (oral, human) = 4111 mg/kg/central nervous system effects/gastrointestinal effects.

SUSPECTED CANCER AGENT: Lithium Carbonate is not found on the following lists: FEDERAL OSHA Z LIST, NTP, IARC, CAL-OSHA and is therefore not considered to be, or suspected to be, a cancer-causing agent by these agencies. IRRITANCY OF PRODUCT: This product is expected to cause irritancy to the skin and eyes.

SECTION 11 TOXICOLOGICAL INFORMATION (Continued)

SENSITIZATION TO THE PRODUCT: Lithium Carbonate can cause allergic dermatitis (cracking and reddening of the skin) upon repeated or prolonged over-exposures

REPRODUCTIVE TOXICITY INFORMATION: Lithium Carbonate is used as a medication for manic-depression. Overexposures to Lithium Carbonate may cause reproductive disorders, based on clinical tests with laboratory animals. Lithium Carbonate may cause fetal harm when administered to a pregnant woman. Children of mothers who received Lithium Carbonate during the first three months of pregnancy have reported in some, but not all, studies to have a slightly increased frequency of malformations of the heart and blood vessels. Even though this risk is low and uncertain, it is strongly recommended that women discontinue lithium therapy during the first three months of pregnancy. Additionally, Lithium is excreted in human milk. Nursing should not be undertaken during lithium therapy except in rare and unusual circumstances.

NOTE! It is important for pregnant women not to be exposed above the exposure limits defined in Section 2 (Composition and Information on Ingredients) during the first trimester, due to the reported teratogenicity of lithium carbonate at high doses.

Mutagenicity: Human mutation data is reported for lithium carbonate, during clinical studies of specific human tissue exposed to relatively high doses.

Embryoloxicity: This product is not reported to produce embryoloxic effects in humans.

Teralogenicity: Clinical studies involving test animals exposed to high doses of Lithium Carbonate indicate teratogenic effects.

Reproductive Toxicity: Clinical studies involving test animals exposed to high doses of Lithium Carbonate indicate adverse reproductive effects.

A mutagen is a chemical which causes permanent changes to genetic material (DNA) such that the changes will propagate through generational lines. An embryotoxin is a chemical which causes damage to a developing embryo (i.e. within the first eight weeks of pregnancy in humans), but the damage does not propagate across generational lines. A teratogen is a chemical which causes damage to a developing fetus, but the damage does not propagate across generational lines. A reproductive toxin is any substance which interferes in any way with the reproductive process...

MEDICAL CONDITIONS AGGRAVATED BY EXPOSURE: Pre-existing respiratory, skin, central nervous system, and kidney conditions can be aggravated by over-exposure to this product.

RECOMMENDATIONS TO PHYSICIANS: Respiratory problems, dermatitis and skin disorders can be aggravated by exposure to this product

BIOLOGICAL EXPOSURE INDICES: Currently there are no Biological Exposure Indices (BEIs) associated with Lithium Carbonate.

Additional detailed toxicology information is presented in the sections below.

LITHIUM SALTS:

ACUTE EXPOSURE: Ingestion of a large dose of lithium salts may cause severe gastroenteritis and effects on the central nervous system, renal function and fluid and electrolyte balance. Symptoms, possibly delayed, may include nausea, vomiting, thirst, anorexia, diarrhea, blurred vision, drowsiness, weakness, tremor, staggering, bradycardia and coma. More unusual reactions may include delirium with EEG changes, action myoclonus, rhabdomyolysis, ECG changes, glycosuria, and allergic erythema. A painful discoloration of the fingers and toes and coldness of the extremities within 1 day of therapeutic use has been reported. In severe cases, death may occur due to renal failure or cardiac or pulmonary complications. Some survivors may have long-lasting or permanent sequelae, mostly of cerebellar nature but, sometimes with peripheral neuropathy or parkinsonism

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CHRONIC EXPOSURE: Repeated or prolonged ingestion of lithium salts may cause symptoms as detailed in acute

TOXICOLOGICAL INFORMATION (Continued)

SECTION 11

Romberg sign, blackout spells, stupor, tinnitus, unconsciousness and coma. Neurologic asymmetry, psychomotor retardation, slurred speech, nystagmus, changes in the EEG and epileptiform seizures may occur. Pseudotumor cerebri (increased intracranial pressure and papilledema) has been reported and may possibly result in enlargement of the blind spot, constriction of visual fields and eventual blindness due to optic atrophy. Photophobia has been reported. Muscular effects may include tremors, ataxia, muscular and reflex hyperimitability with fasiculations, twitching and spastic or

choreo-athetotic movements, cogwheel rigidity, parkinsonism and dystonia. Two cases involving severe generalized sensorimotor peripheral neuropathy have been reported. ECG changes, cardiac arrhythmias, hypotension, peripheral

hypernatremia and hyperosmolality, sodium depletion, polyuria, glycosuria, oliguria, anuria, and azotemia are possible. Morphologic changes with glomerular and interstitial fibrosis and nephron atrophy have been reported. However, a causal relationship has not been established. Dermatologic effects may include cutaneous hyperalgesia or anesthesia, xerosis cutis, chronic folliculitis, generalized pruritus with or without rash, development or exacerbation of acne or psoriasis, cutaneous ulcers and alopecia. Hyper- or hypothermia, weight gain, edema of the ankles and wrists and sexual dysfunction have been reported. Death may occur due to renal failure, brain damage or pulmonary complications.

ingestion. In addition, a metallic taste, dry mouth, excessive thirst, abdominal pain and incontinence of urine and feces may occur. Nervous system effects may include a dazed feeling, confusion, giddiness, mental lapses, dyspraxia, drowsiness, vertigo, headache, apathy, restlessness, anxiety, some suppression of the rem phases of sleep, positive

circulatory collapse, and interstitial myocarditis are possible. Leukocytosis is fairly common. Endocrine effects may include disturbed iodine metabolism, stimulation of antithyroidal auto-antibodies, hypothyroidism with myxedema, or rarely hyperthyroidism. Osteoporosis, an increase in serum total calcium, ionized calcium and parathyroid hormone and independently functioning parathyroid adenomas have been reported. Transitory nephrotic syndrome and acquired nephrogenic diabetes insipidus may occur. Transient hyperglycemia, lowered urinary concentrating ability leading to

Lithium readily crosses the placental barrier and is excreted in breast milk. The use of lithium during pregnancy has been associated with neonatal goiter, cardiac anomalies, especially ebstein's, central nervous system depression and hypotonia. Marked functional and structural changes in the kidneys of newborn rats exposed to lithium via their mother's milk have been reported. Adverse effects on nidation in rats and embryo viability in mice have been attributed to lithium, as have teratogenicity in submammalian species and cleft palates in mice. However, studies in rats, rabbits

and monkeys have shown no evidence of lithium-induced developmental defects. Leukemia has been reported during lithium treatment. However, an epidemiologic study involving a population of 173,000 persons yielded negative results

ANTIDOTE: The following antidote has been recommended. However, the decision as to whether the severity of poisoning requires administration of any antidote and actual dose required should be made by qualified medical personnel.

LITHIUM/LITHIUM SALT POISONING:

NOTE TO PHYSICIAN

1) In single ingestion episodes, give syrup of ipecac and/or perform gastric lavage if productive vomiting has not already occurred. 2) Fluid and electrolyte replacement for the correction of dehydration and acid-base imbalances. Overhydration may precipitate pulmonary edema. 3) Infusion of urea or mannitol, alkalinization of the urine and, and

aminophylline increase lithium excretion in patients with good renal function. 4) Extracorporeal or peritoneal hemodialysis to decrease lithium levels and control uremia in severe intoxications. If a massive overdose is known with certainty to have been ingested, it may be prudent to institute these measures even in the absence of positive clinical findings because of severe delayed toxicity. 5) Diazepam for the suppression of abnormal motor activity. 6) Support treatment for central nervous system depression. 7) Frequent electrocardiograms to assess cardiac status. (Gosselin,

Smith, Hodge - Clinical Toxicology of Commercial Products, Fifth Edition). Activated charcoal does not bind lithium effectively and is not useful in isolated lithium toxicity. (Groleau, Lithium Toxicity, Emergency Medicine Clinics of North America, Volume 12, Number 2, May, 1994). Raising the sodium intake does not increase lithium clearance (Thomsen, K. Renal lithium elimination in man and active treatment of lithium poisoning. Acta Psychiatr. Scand., Suppl. No. 207:83-84,1969).

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ECOLOGICAL INFORMATION SECTION 12

ENVIRONMENTAL STABILITY: Lithium Carbonate is stable in the environment.

EFFECT OF MATERIAL ON PLANTS or ANIMALS: The effects on exposed animals would be primarily irritation

of contaminated tissue (see Section 11, Toxicological Information). The main effect on plants would be the increase in

salinity of contaminated soils if large volumes of this product are released. As with all chemicals, work practices should

be aimed at minimizing environmental releases.

EFFECT OF CHEMICAL ON AOUATIC LIFE: Releases of large quantities of this product can be detrimental to an aquatic environment, by altering the salinity of a body of water. As with all chemicals, work practices should be aimed

at minimizing environmental releases. ACUTE AOUATIC TOXICITY: No data available.

DEGRADABILITY: No data available. LOG BIOCONCENTRATION FACTOR (BCF): No data available.

LOG OCTANOL/WATER PARTITION COEFFICIENT: No data available.

SECTION 13 DISPOSAL CONSIDERATIONS

PREPARING WASTES FOR DISPOSAL: Waste disposal must be in accordance with appropriate Federal, State, and local regulations. This chemical, if unaltered by use, may be disposed of by treatment at a permitted facility or as advised by your local solid waste regulatory authority.

EPA WASTE NUMBER: Not applicable to the product.

SECTION 14 TRANSPORT INFORMATION

THIS MATERIAL IS NOT HAZARDOUS AS DEFINED BY 49 CFR 172.101 BY THE U.S. DEPARTMENT OF

TRANSPORTATION.

PROPER SHIPPING NAME: Not applicable. **HAZARD CLASS NUMBER and DESCRIPTION:** Not applicable.

UNIDENTIFICATION NUMBER: Not applicable. PACKING GROUP: Not applicable. DOT LABEL(S) REQUIRED: Not applicable.

NORTH AMERICAN EMERGENCY RESPONSE GUIDEBOOK NUMBER, 1996: Not applicable.

MARINE POLLUTANT: No component of this product is designated as a DOT Marine Pollutant (49 CFR 172.101,

Appendix B). TRANSPORT CANADA TRANSPORTATION OF DANGEROUS GOODS REGULATIONS: THIS MATERIAL IS NOT CONSIDERED AS DANGEROUS GOODS.

EMERGENCY RESPONSE CONTACT FOR AN INCIDENT DURING TRANSPORTATION: CHEMTREC 1-800-424-9300 or 1-703-527-3887

SECTION 15 REGULATORY INFORMATION

SARA REPORTING REQUIREMENTS: Lithium Carbonate is subject to the reporting requirements of the Comprehensive Environmental Response, Compensation, and Liability Act and Sections 302, 304 and 313 of Title III

of the Superfund Amendments and Reauthorization Act., as follows:

CERCLA SECTION 103 (40 CFR 302.4): No

No

SARA SECTION 302 (40 CFR 355.30):

SARA SECTION 304 (40 CFR 355.40): No Ycs

SARA SECTION 313 (40 CFR 372.65):

not induces vomiting. Seek medical attention. IN CASE OF FIRE: Use water fog, dry chemical, CO2, or "alcohol" foam. IN CASE OF SPILL: Sweep-up or vacuum spilled material. Place in a suitable container. Consult Material Safety Data Sheet before use.

WARNING! This product contains a chemical known to the State of California to cause birth defects or other reproductive harm.

TARGET ORGANS: Eyes, skin,. (via inhalation or ingestion: central nervous system, kidneys).

WHMIS SYMBOLS: Not applicable.

OTHER INFORMATION SECTION 16

REVISIONS - Corporation name change

regulations.

PREPARED BY:

The information in this Material Safety Data Sheet is based on data that Chemetall Foote Corporation believes to be reliable as of the MSDS's

date of revision. Chemetall Foote Corporation makes no warranty or representation of any kind that the MSDS does not contain errors. The data in this MSDS relates only to the specific material designated herein and does not relate to use in combination with any other material or in any process. It is intended for use by persons having technical skill and at their own discretion and risk. Since conditions of use are outside the

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CHEMICAL SAFETY ASSOCIATES, Inc.

DEFINITIONS OF TERMS

A large number of abbreviations and acronyms appear on a MSDS. Some of these which are commonly used include the following:

CAS #: This is the Chemical Abstract Service Number which uniquely identifies each constituent. It is used for computer-related searching. EXPOSURE LIMITS IN AIR:

ACGIH - American Conference of Governmental Industrial Hygienists, a professional association which establishes exposure limits.

SECTION 16 OTHER INFORMATION (Continued)

TLV - Threshold Limit Value - an airborne concentration of a substance which represents conditions under which it is generally believed that nearly all workers may be repeatedly exposed without adverse effect. The duration must be considered, including the 8-hour Time Weighted Average (TWA), the 15-minute Short Term Exposure Limit, and the instantaneous Ceiling Level. Skin adsorption effects must also be considered.

OSHA - U.S. Occupational Safety and Health Administration.

PEL - Permissible Exposure Limit - this exposure value means exactly the same as a TLV, except that it is enforceable by OSHA. The IDLH - Immediately Dangerous to Life and Health level represents a concentration from which one can escape within 30-minutes without suffering escape-preventing or permanent injury. The DFG - MAK is the Republic of Germany's Maximum Exposure Level, similar to the U.S. PEL. NIOSH is the National Institute of Occupational Safety and Health, which is the research arm of the U.S. Occupational Safety and Health Administration (OSHA). NIOSH issues exposure guidelines called Recommended Exposure Levels (RELs). When no exposure guidelines are established, an entry of NE is made for reference.

FLAMMABILITY LIMITS IN AIR: Much of the information related to fire and explosion is derived from the National Fire Protection Association (NFPA). <u>LEL</u> - the lowest percent of vapor in air, by volume, that will explode or ignite in the presence of an ignition source. <u>UFL</u> - the highest percent of vapor in air, by volume, that will explode or ignite in the presence of an ignition source.

TOXICOLOGICAL INFORMATION

Possible health hazards as derived from human data, animal studies, or from the results of studies with similar compounds are presented. Definitions of some terms used in this section are: LD₅₀ - Lethal Dose (solids & liquids) which kills 50% of the exposed animals; LC₅₀ - Lethal Concentration (gases) which kills 50% of the exposed animals; ppm concentration expressed in parts of material per million parts of air or water; mg/m² concentration expressed in weight of substance per volume of air; mg/kg quantity of material, by weight, administered to a test subject, based on their body weight in kg. Data from several sources are used to evaluate the cancer-causing potential of the material. The sources are: IARC - the International Agency for Research on Cancer; NTP - the National Toxicology Program. RTECS - the Registry of Toxic Effects of Chemical Substances, OSHA and CAL/OSHA. IARC and NTP rate chemicals on a scale of decreasing potential to cause human cancer with rankings from 1 to 4. Subrankings (2A, 2B, etc.) are also used. Other measures of toxicity include TDLo, the lowest dose to cause a symptom and TCLo the lowest concentration to cause a symptom; TDo, LDLo, and LDo, or TC, TCo, LCLo, and LCo, the lowest dose (or concentration) to cause death. BEI - Biological Exposure Indices, represent the levels of determinants which are most likely to be observed in specimens collected from a healthy worker who has been exposed to chemicals to the same extent as a worker with inhalation exposure to the TLV.

This section explains the impact of various laws and regulations on the material. EPA is the U.S. Environmental Protection Agency. WHMIS is the Canadian Workplace Hazard information System. DOT and CTC are the U.S. Department of Transportation and the Canadian Transportation Commission, respectively. These are: Superfund Amendments and Reauthorization Act (SARA); the Toxic Substance Control Act (TSCA);

Marine Pollutant status according to the DOT; California's Safe Drinking Water Act (Proposition 65); the Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA or Superfund); and various state regulations. This section also includes information on the precautionary warnings which appear on the materials package label.

GRAPHICAL REPRESENTATION OF HAZARDS

HAZARDOUS MATERIAL INFORMATION SYSTEM NATIONAL FIRE PROTECTION SYSTEM RATING RATING



